

Effects of Ultrasound Treatments on Antioxidants Content of Cider, Enriched Previously with Natural Extracts

BORIS BREZAN^{1*}, CARMEN LILIANA BADARAU², ALEXANDRU WOINAROSCHY^{1*}, VASILE PADUREANU²

¹ University Politehnica of Bucharest, Faculty of Applied Chemistry and Materials Science, 1-7 Polizu Str., 011061, Bucharest, Romania

²Transilvania University of Brasov, Food and Tourism Faculty, 148 Castelului, 500014, Brasov, Romania

Abstract. The cider obtained in laboratory conditions was enriched in valuable and natural bio compounds using several extracts (obtained from blueberry juice and an colorant obtained from black carrot "Black Carrot HC Red-Blue Colored Concentrate" in 2% citric acid solution) and ultrasounds treatments. Compared with the untreated cider, all the cider variants enriched with both types of extracts had higher values for total polyphenols and flavonoids content. This increase in values, from one measurement to another is due to the increase in the amplitude of the ultrasound in the propagation medium (cider + liquid extract). This reveals the importance of increasing the amplitude, in the efficiency of extraction of anthocyanins, demonstrating that a similar application time, but at different amplitude, influences the content of polyphenols and flavonoids, respectively. To assess the antioxidant activity of the cider variants tested in the experimental study, it was estimated the diphenylpicrylhydrazyl (DPPH) free inhibition percentage (DPPH inhibition) by the enriched cranberry extracts obtained from the forest fruits (blueberries) in 2% citric acid, respectively by those obtained by using black carrot juice (0.03% concentration).

Keywords: antioxidant compounds, cider, diphenylpicrylhydrazyl (DPPH)

1.Introduction

Over time, especially in recent years, the demand for minimally processed foods has led to significant changes in the techniques used, as some of them applied under certain critical conditions decreased the nutritional value of the product, and to the same extent its bioavailability, due to changes in physical or chemical nature, ultimately leading to reduced organoleptic acceptability.

The use of ultrasound is classified as a non-thermal processing method, with numerous advantages deriving from it, being in the same time one of the quickest techniques that have been developed to minimize processing, improve quality and protect food safety primarily [1,2]. All of these things, besides many alternatives they can offer, compared to conventional methods or techniques, since they can be used in the food industry from analysis to product modification [3, 4].

The paper wishes to reveal the importance of their application for the purpose of conducting a study on the potential for enrichment of cider by releasing phenolic compounds and anthocyanins, in the end.

The application of ultrasound in the case of liquids, as in the case of the present work, is based on the phenomenon of cavitation, which involves the formation, development and implosion of bubbles along with the generation and propagation of vibrations in the mass it crosses. The implosion, finally, causes mechanical (turbulent, shear), thermal and also chemical effects, with the generation of heat and pressure at the same time [1].

The work done for the experimental study focused on the following aspects:

applying different ultrasound treatments (e.g. different amplitudes) in the case of improved cider samples by adding rich extract of antioxidant compounds initially (e.g. different extract volumes),

^{*}email: borisalexid@yahoo.com; awligi@yahoo.com



- Physical *Diphenylpicrylhydrazyl* chemistry tests to get acquainted with the modification of the antioxidant content,

Abbreviation: TPC - total polyphenol content; TFC – total flavonoids content; DPPH – Diphenylpicrylhydrazyl

2.Material and methods

Collection and preparation of plant material

The plant sources used to obtain the analyzed extracts to determine the content of polyphenols and flavonoids, respectively the assessment of antioxidant activity, were cranberries from Duke variety, purchased from the supermarket, culture origin, in a fresh state.

Extraction

Prior to extraction, the samples were cleaned and then wetted. A mass of freshly moistened samples (50 g) was added to the amount of solvent initially dosed (200 mL) in the sample mixing vessels, the mixture being subsequently stored at a temperature of 20°C for 24 h. After this interval, the sample was subsequently shaken, using a Vortex shaker.

-Regarding the used solvents, extraction conditions that were applied to the plant source, makes direct reference to citric acid 2%.

After the extraction was complete, the samples were centrifuged at 10,000 rpm, 15 min.

All experiments were performed in three repetitions, and the results are expressed as mean value \pm standard deviation.

The main operations of the extraction scheme for polyphenols and flavonoids are shown in the following:

-samples homogenizing and stirring, subsequently added over the solvent,

-filtration, subsequently to 24 h storage of mixture at 20°C,

-centrifugation, in order to collect the supernatant.

Tested variants

In order to carry out the experimental part, it was proceeded to use the cider obtained previously, in order to enrich it with a liquid extract concentrated in antioxidant compounds, obtained in laboratory conditions similar to the previous ones.

Various variants were tested, according to the following table:

Table 1. Cider variants with cranberry extract					
Coding	Amplitude	Vol. of cider	Vol. of extract [mL]		
	[%]	[mL]			
AM	-	50	2.5		
A1	20	100	5		
A2	30	100	5		
A3	40	100	5		
A4	50	100	5		

 \overline{AM} = witness tip sample; A1...A4 = 1 st variant using cranberries extract...4th variant using cranberries extract etc.

Also, it was tested a colorant used at industrial level, named in international language "Colored Concentrate Black Carrot HC Red-Blue", manufactured by Dohler GMBH company in Darmstadt, Germany. Regarding the tested variants, table 2 contains details regarding cider additions



Coding	Amplitude	Vol. of cider	Vol. of extract
	[%]	[mL]	[mL]
MM	-	50	2.5
M1	20	100	5
M2	30	100	5
M3	40	100	5
M4	50	100	5

Table 2. Cider variants with black carrot concentrated*

*0.03% concentration

Determination of Total Phenolic Content (TPC)

Determination of total polyphenols content from extracts, involved:

- crushing the material and then, approx. 1g of the product under test to be homogenized with 10 mL 96% ethanol, shaken vigorously for 2-3 min (Vortex) and filtered through a strained filter; then, 0.5 mL extract and 1.5 mL Folin Denis reagent (1:10) are introduced into a tube,

- standing for 5 min and then adding 2 mL 7.5% sodium carbonate solution,
- leaving to stand for 90 min and then read absorbance at 725,
- using the calibration curve the total polyphenol content was estimated.

For the determination of the calibration curve, necessary to estimate the content of polyphenols in samples, was used gallic acid (Roth, art no. 2699.1, Germany).

Determination of Total Flavonoids Content (TFC)

In order to determine the total flavonoids content, the material is crushed (if applicable) and then, 1 g of the product under test is homogenized with 10 mL of acidified ethanol (with 1% HCl), shaken vigorously for 2-3 min and filtered through a filter. For an extraction as advanced as possible, the operation can be repeated 2-3 times.

A 0.5 mL extract is introduced into a tube, over which 2mL distilled water is added and then 150 μ L5% NaNO2. Leave to rest for 6 min. Then add 150 microliters of 10% AlCl₃ solution, leave to rest for 6 min, add 2mL of 2NaOH solution and after 15 min of rest read absorbance at 510nm (Spectrophotometer DR2800 Hach Lange, USA), relative to distilled water. Using the calibration curve, the total flavonoid content is estimated.

Determination of the calibration curve required to estimate the flavonoids content of the samples, assumed the use of quercetin dihydrat.

DPPH Activity

An adapted method was used after Yi et all. (Yi Z., YU Y., Liang Y, Zeng B.2008, p. 597-603). After filtering the samples, in the alveoli of an ELISA plate with 96 wells, it was distributed using a pipette of 20 microlitres, distilled water and then 20 microlitres sample (filtered beforehand). There, were then added 200 microlitres of DPPH 0, 3mM solution in each alveole. For controls, equal volumes of distilled water were used instead of the sample. The plates were incubated in the dark for 30 min. The absorbance was read at 515nm wavelength using the TecanSunRise reader (Magellan software).

The percentage of DPPH radicals inhibition was calculated using the formula:

% DPPH inhibition = [(A control – A sample)/A control] x 100 (%)

where A represents the absorbance read at 515nm [5]

3.Results and discussions

The values obtained represent the average of the readings, for absorbent samples at a wave length of 725nm. The figures vary quite a lot with each measurement, the results depending on the amplitude



that was used. Maximum phenols were found in formulation consisting cider enriched with aqueous extract, and subject to a 50% amplitude treatment. Table 3 describes the TPC for all tests carried out for cider enriched both with citric acid 2% and black carrot concentrated.

		Sample volume	Solvent	ent in analyzed of Average		
SOLVENT	Sample	[mL]	[mL]	Abs 725nm	TPC, mg/mL	TPC, mgAGE/100g
	AM	1	10	0.753	0.320	319.90
	A1	1	10	0.824	0.353	353.36
WATER CITRIC ACID 2%	A2	1	10	0.861	0.370	370.48
	A3	1	10	0.939	0.407	407.39
	A4	1	10	1.940	0.879	878.89
	MM	1	10	0.689	0.290	289.75
0.03% BLACK CARROT CONCENTRATED ADDITION	M1	1	10	0.768	0.327	326.82
	M2	1	10	0.743	0.315	315.04
	M3	1	10	0.903	0.390	390.43
	M4	1	10	0.627	0.260	260.22

On the other hand, table 4 compress the results that reveal the total content of flavonoids for the same variants of cider. We can state that we have obtained higher values from one measurement to another, for each of the ciders enriched with both types of extracts.

		Sample vol.	Solvent vol.	Average			
SOLVENT	Sample	[mL]	[mL]	Abs 510nm	TFC, mg/mL	TFC, mg/100g	
	AM	1	10	0.159	0.013	12.94	
	A1	1	10	0.182	0.015	14.91	
CITRIC ACID 2%	A2	1	10	0.331	0.028	27.65	
	A3	1	10	0.421	0.035	35.36	
	A4	1	10	0.573	0.048	48.36	
0.03% BLACK CARROT CONCENTRAT ED ADDITION	MM	1	10	0.135	0.011	10.88	
	M1	1	10	0.230	0.019	18.96	
	M2	1	10	0.268	0.022	22.22	
	M3	1	10	0.326	0.027	27.19	
	M4	1	10	0.193	0.016	15.85	

Table 4. Total flavonoids content in analyzed ciders



This increase in values, from one measurement to another, is due to the modification (growth) of the ultrasound amplitude in the propagation medium (cider plus liquid extract). This reveals the importance of increasing amplitude in streamlining the extraction of anthocyanin compounds, demonstrating that a similar application time, but at a different amplitude, influences the content in phenols, respectively flavonoids. The best results in the extraction of phenols and flavonoids from the ciders enriched with cranberry extract were obtained using a amplitude of 50% in both cases. In the same time, in the case of extraction of phenols and flavonoids from the ciders enriched with black carrot extract, they were obtained using a amplitude of 40% in both cases.

For the appreciation of the antioxidant activity of the cider variants tested in the experimental study, it was used as reagent, the DPPH. More specifically, it was estimated the percentage of inhibition of DPPH free radicals (% DPPH inhibition) by ciders enriched with extracts obtained from berries (blueberries) in 2% citric acid, respectively with those obtained by using the Black carrot concentrate.

Table 5 shows the average of the values for repetitions of readings at wavelength of 515 nm, respectively the percentage of DPPH inhibition, for berries, but also for the black carrot concentrate.

	Sample	Values o	Values of Asssnm	
SOLVENT		AVERAGE	% inh DPPH	
	AM	0.373	72.83	
	A1	0.419	69.46	
CITRIC ACID 2%	A2	0.366	73.32	
	A3	0.346	74.82	
	A4	0.394	71.28	
	MM	0.307	77.61	
	M1	0.306	77.70	
0.03% BLACK CARROT CONCENTRATED ADDITION	M2	0.450	67.22	
	M3	0.315	77.02	
	M4	0.373	72.83	

4.Conclusions

The content of the works incorporates the potential application of ultrasounds in order to improve the chemical profile of the cider (direct reference to the compounds of the antioxidant nature) and the mechanisms of the phenomenon, with direct reporting to the cider.

The study reveals importance of ultrasounds application for increasing content in such compounds, considering its application to different amplitudes. The explanation is the partitionation of the molecules present in the initial sample, followed by the intensification of the activity of these compounds in the end, as a result of their division (e.g. shear).

A linear increase in the temperature is observed in both variants of cider, which is due to the increase of the amplitude of ultrasounds. Once we increase the amplitude, it occurs the increase in the temperature of the ultrasonicated environment, and figure 1 showing the changes in this direction. That's why, side effects of their use, on other cider compounds, can be taken into account in future studies, as long as temperature in range of 50 - 60 $^{\circ}$ C were achieved.



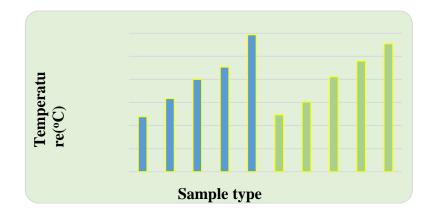


Figure 1. Temperature modification during ultrasonication

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